

Springs Protection Assessment Using Nitrate Isotope Analysis

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Springs Protection Assessment Using Nitrate Isotope Analysis OUTLINE



- Why is nitrogen so important?
 - Nitrogen and the Springs and Aquifer Protection Act
 - Nitrogen Sources
- Dual Nitrate Isotopes
 - What they are
 - How they work
 - Data Limitations
 - Practical guidelines for sampling
- Case Study: Nitrate sources to Wekiwa spring
 - Sources of nitrate to spring
 - Mixing models
 - Outcomes: Identification of hotspots, Management Practices



Springs Protection Assessment Using Nitrate Isotope Analysis WHY IS NITROGEN SO IMPORTANT?



Nitrogen Paradox: One of the 5 elements necessary for life, BUT more than 99% occurs as triple bonded N₂, which cannot be used by organisms.

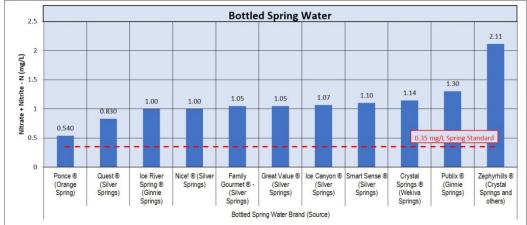
Breaking the bond requires lots of energy

- High temperature processes (lightning)
- Nitrogen-fixing microbes

Anthropogenic inputs have caused concentrations to rise to harmful levels

- High nitrate can cause eutrophication
- In Florida, eutrophication \rightarrow impaired springs
- 90% Florida's drinking water from GW
 - Springs 'window' into health of GW
 - Threaten recreational /economic value of springs





Florida Springs Institute 2018

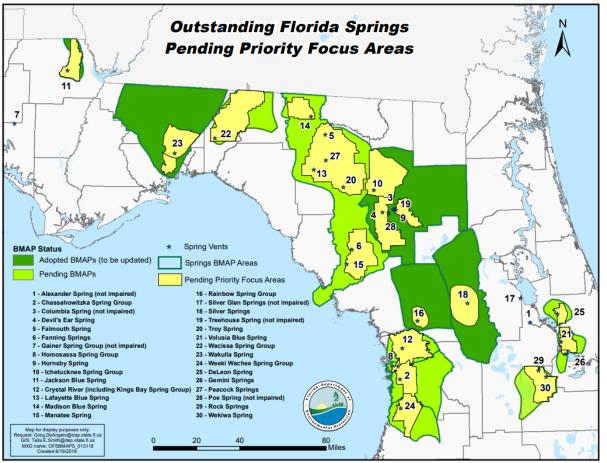


Springs Protection Assessment Using Nitrate Isotope Analysis SPRINGS AND AQUIFER PROTECTION ACT 2016



FDEP adopted 13 restoration plans in 2018 to address 24 Nitrogen impaired Outstanding Florida Springs.

- Delineate priority focus areas;
- Prioritize lists of restoration projects which include estimates of nutrient load reduction;
- Project milestones to achieve water quality restoration targets in 20 years;
- Estimated nutrient pollutant loads, allocated to each source or category of sources; and
- Completed remediation plans for onsite sewage treatment and disposal systems (OSTDS) where septic loading accounts for at least 20 percent of the estimated nutrient input.



http://www.floridahealth.gov/environmental-health/onsitesewage/research/_documents/ofs_bmaps_and_pfas.pdf





Fertilizers contain varying proportions of nitrogen, nitrogen species, and degrees of slow-release nitrogen.

- Lawn or crop-applied
- UF/IFAS estimates 1-55% of fertilizer nitrogen applied on Florida lawns leaches to groundwater (Shaddox and Unruh 2018)



-Adobe Stock Image



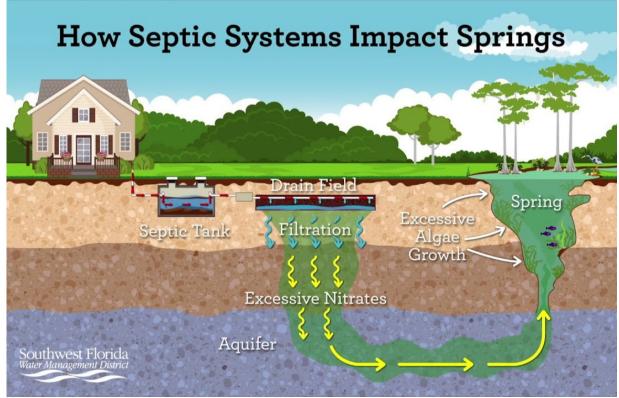


Springs Protection Assessment Using Nitrate Isotope Analysis COMMON NITRATE SOURCES



Human (Wastewater / Septic Leachate) and Animal Wastes contain fixed nitrogen that can reach springs.

- Urine is mineralized to ammonia, then oxidized to nitrate.
- Nitrate is very soluble; when it enters the aquifer, it moves with water.



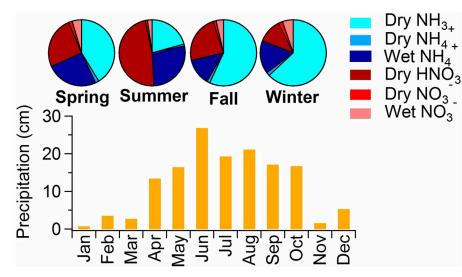


Springs Protection Assessment Using Nitrate Isotope Analysis COMMON NITRATE SOURCES

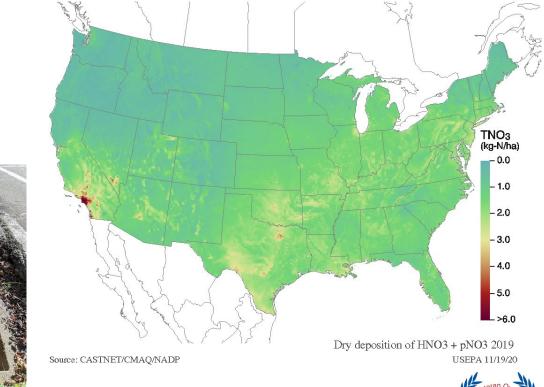


Atmospheric Nitrogen accumulates in the atmosphere from powerplants, vehicles, and industrial emissions.

• Nitrogen is deposited through wet and dry deposition.



Pie charts of seasonal N deposition species pathways (Upper) and total monthly measured precipitation (Lower) in Florida area (Li et al. 2016)





Springs Protection Assessment Using Nitrate Isotope Analysis COMMON NITRATE SOURCES

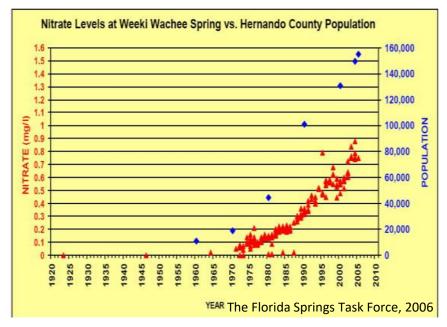
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1/1/1985



Nitrogen is naturally "fixed" in the soil, but natural / background (non-human input) groundwater nitrate concentrations are very low.



Nitrate-N concentrations at Weeki Wachee Spring (red triangles) and Population (blue diamonds) Hernando County. Nitrate-N concentrations in Juniper (conservation lands) and Lithia (mix of agriculture, urban/ residential).

1/1/1995

Date

1/1/1990

Nitrate Concentrations in Juniper and Lithia Springs

[Juniper - NO3-T; Lithia - NO3-D]

The Florida Springs Task Force, 2006

1/1/2000

1/1/2005

".....Throughout the karst areas of north and central Florida where artesian springs are common, groundwater nitrate nitrogen concentrations have increased from a normal background of *less than 0.02 parts per million* to widespread concentrations over 1.0 parts per million (a fifty-fold increase)..." (from Florida, Land of 1,000 Springs, Howard T. Odum Florida Springs Institute, http://www.lake.wateratlas.usf.edu/upload/docum ents/Springs-Facts-FSI-012414.pdf



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- Mass differences cause different reaction rates or "fractionations" during chemical and biological reactions.
- Fractionations allow fingerprinting of sources and reactions.
- δ (Delta) Notation: $\delta_{\text{Sample}} = ((R_{\text{sample}} R_{\text{standard}})/R_{\text{standard}})*1000$

where "R" is the ratio of the heavy to light isotope in the sample or standard. A positive " δ " value means the sample is heavier than a standard, negative " δ " values indicates the sample is lighter than the standard.





Analytical advances have transformed our capacity for analyzing nitrate

isotopes.

Sample mass requirements decreased by 3 orders of magnitude





Denitrifier Method 10-20 nmol NO₃⁻



Slide courtesy of Dr. Emily Elliott, University of Pittsburgh

AgNO₃ Method

100-200 µmol NO₃⁻



Analytical advances have **transformed** our capacity for analyzing nitrate isotopes.





https://syringefilter.com/



Springs Protection Assessment Using Nitrate Isotope Analysis DUAL NITRATE ISOTOPES



Analytical advances have transformed our capacity for analyzing nitrate

isotopes.



Pseudomonas aureofaciens

Photo courtesy of "Microbe wiki"

 $NO_3^- \rightarrow NO_2^- \rightarrow N_2O \prec$

δ¹⁵N (Sigman et al., 2001)

δ¹⁸O (Casciotti et al., 2002)

∆¹⁷O (*Kaiser et al., 2007*)

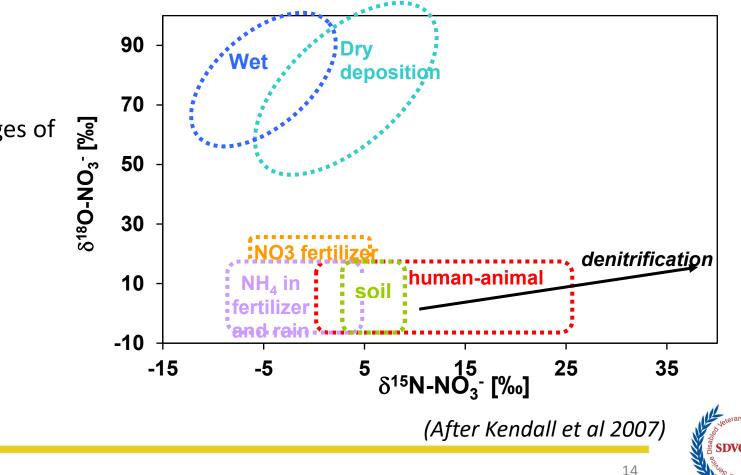
Slide courtesy of Dr. Emily Elliott, University of Pittsburgh





Nitrate isotopes can provide information about what sources contribute to observed nitrate.

- Nitrate sources show distinct ranges of reported values.
- Some overlapping signatures





Denitrification Reaction Sequence
$$NO_{2}^{-} \rightarrow NO_{2}^{-} \rightarrow NO_{2}^{$$

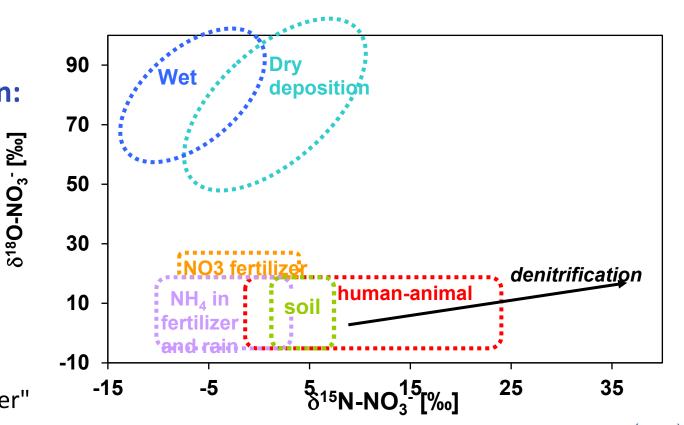
Nitrate Nitrite Nitric Oxide Nitrous Oxide Nitrogen Gas

Conditions that promote denitrification:

- Presence of denitrifying bacteria
- Low dissolved oxygen
- Presence of electron donor (carbon)

Denitrification enriches δ^{15} N & δ^{18} O

- Lighter isotopes are preferentially used by bacteria
- Remaining nitrate pool is isotopically "heavier"

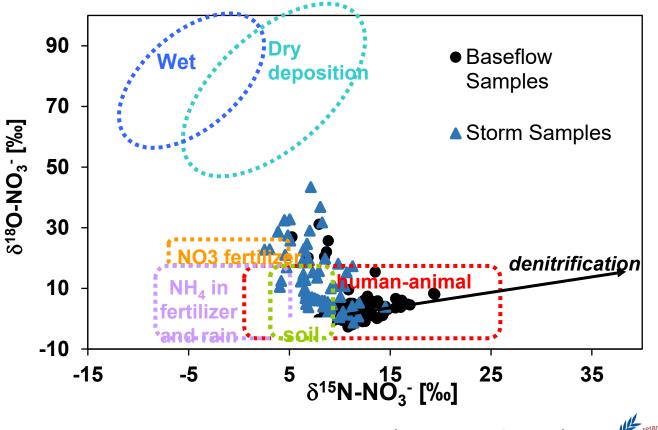






Nitrate isotopes can provide information about what sources contribute to observed nitrate.

- Example Samples from Pittsburgh, PA.
- Surface waters plotted in the dualisotope space.
- Baseflow, (for this water source) influenced by sewage.
- Mixes with atmospheric deposition during storms.
- Denitrification



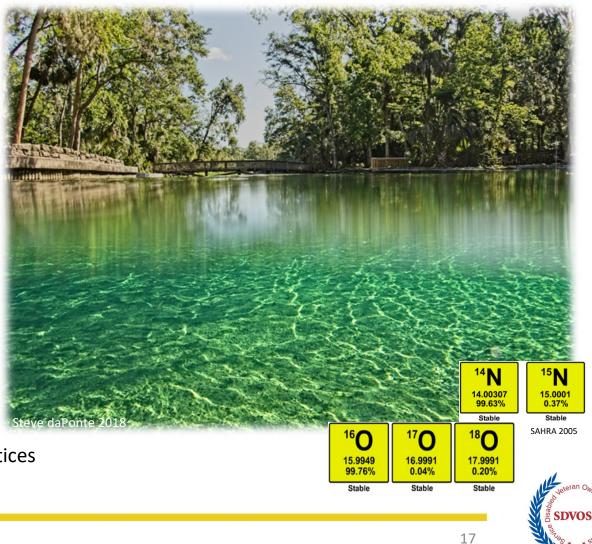
(Divers et al 2014)



Springs Protection Assessment Using Nitrate Isotope Analysis OUTLINE



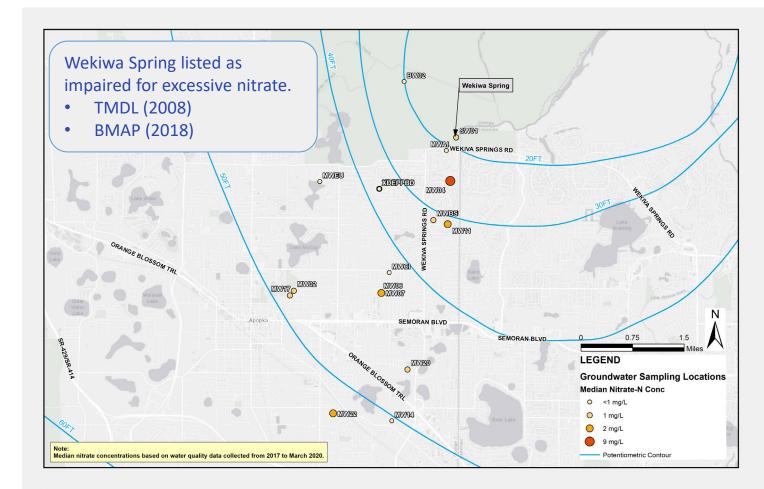
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Is nitrate from fertilizer a significant nutrient source to Wekiwa Spring?

- Water quality data collected quarterly for 3+ years from 22 sampling locations
- Isotopic data collected for sampling locations with higher nitrate concentrations
- Does the data support restrictions on fertilizer practices?







Land use in Wekiwa Priority Focus Area:

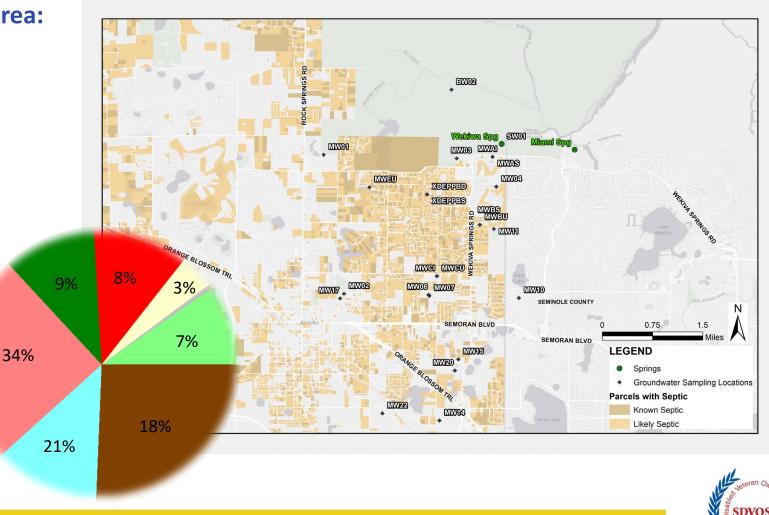
- 34% Urban and Built-Up
- 21% Water
- 18% Wetlands

Septic within Wekiwa Priority Focus

Area:

 Study wells located on parcels connected to sewer

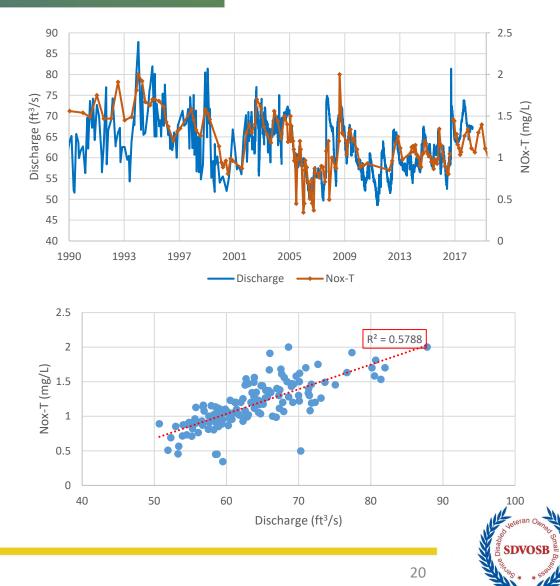
> Agriculture Barren Land Rangeland Transportation, Communication, and Utilities Upland Forest Urban and Built-Up Water Wetlands





Nitrate reaching Wekiwa Springs via groundwater is dependent on a mix of factors:

- Source
 - Continuous = leaky septic
 - Intermittent = fertilizer application
- Groundwater travel times
 - Longer travel time = more denitrification potential
- Hydrogeologic conditions
 - Infiltration rate
 - Conditions to promote or inhibit denitrification

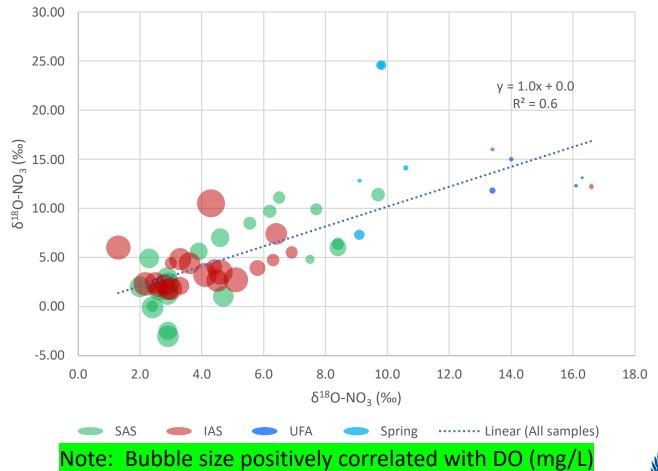




Data suggests mixing of groundwater originating from sources with various nitrogen

loading and/or travel times.

- Samples from wells across Wekiwa Priority Focus area.
 - Spring water from Wekiwa Spring.
 - Wells from Surficial (n=4), Intermediate(n=6) and Upper Floridan (n=1) Aquifers.
- Dissolved oxygen (DO) decreases along denitrification trajectory.

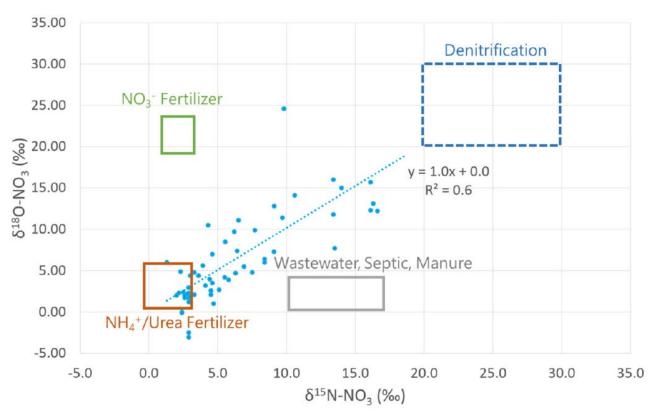






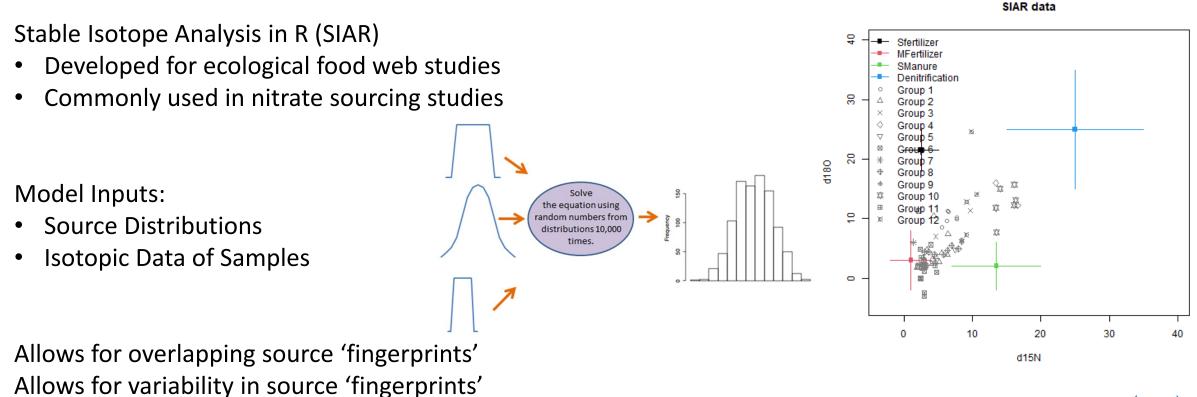
Nitrate Isotope values indicate strong fertilizer signal and denitrification.

- Potential mixing with wastewater.
- Denitrification in action.
 - Albertin et al. (2012) calculated a 43% decline in nitrate concentrations (with increases in δ¹⁵N, δ¹⁸O) for Wekiva Spring from 2005 to 2008.
- Mixing models can help determine proportion of nitrate originating from each source.





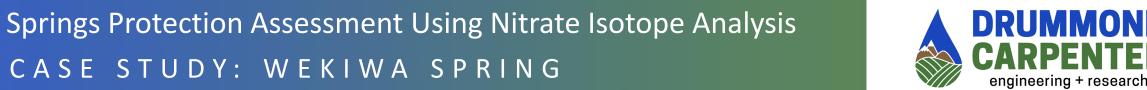
Bayesian Mixing Models can help untangle the proportion of nitrate by source and gain insight into biological processing.





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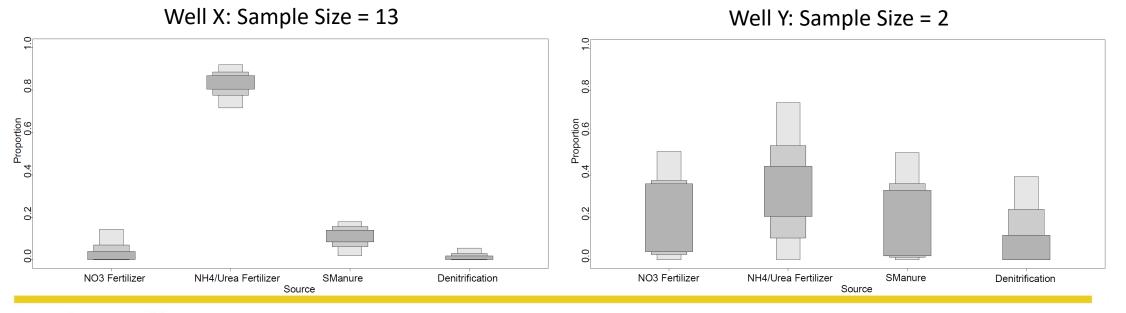
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Mixing model results provide relative contributions of each source to total nitrate in each well sample set.

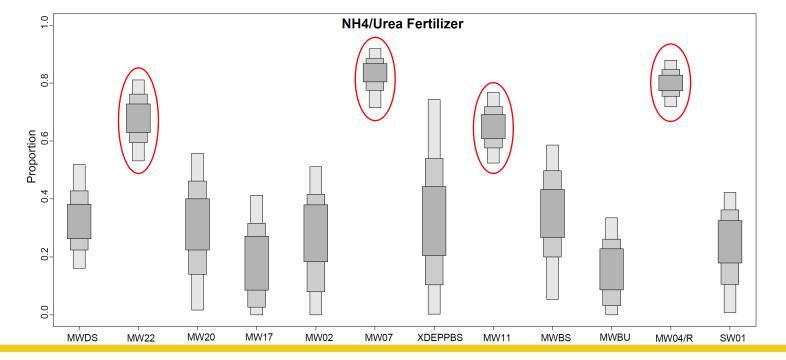
- Mixing model results commonly viewed as boxplots.
 - Allows for visualization of uncertainty.
- Larger sample size produces more certainty in model results.





Mixing model results provide relative contributions of each source to total nitrate in each well sample set.

- Strong fertilizer signature across all sample locations.
- Wells with >60% attributed to fertilizer had average nitrate concentrations > 2 mg/L.

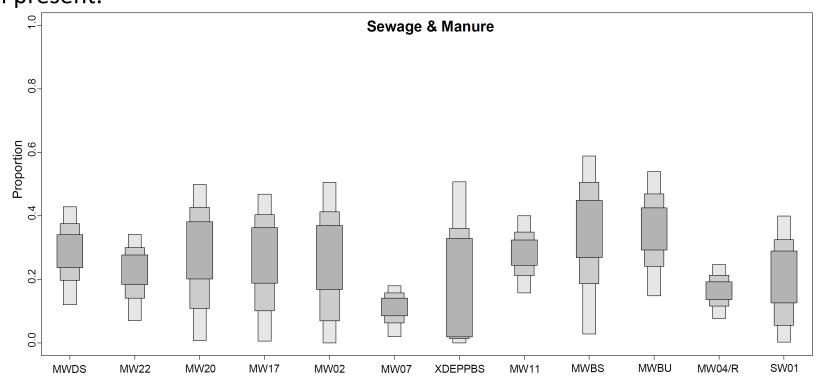






Mixing model results provide relative contributions of each source to total nitrate in each well sample set.

• Sewage signal present.

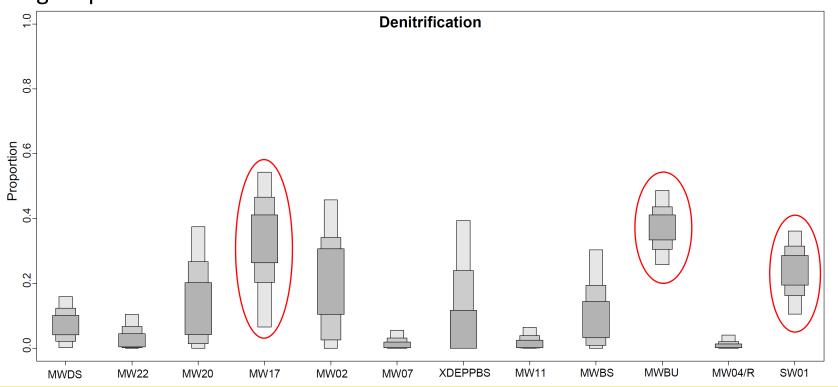






Mixing model results provide relative contributions of each source to total nitrate in each well sample set.

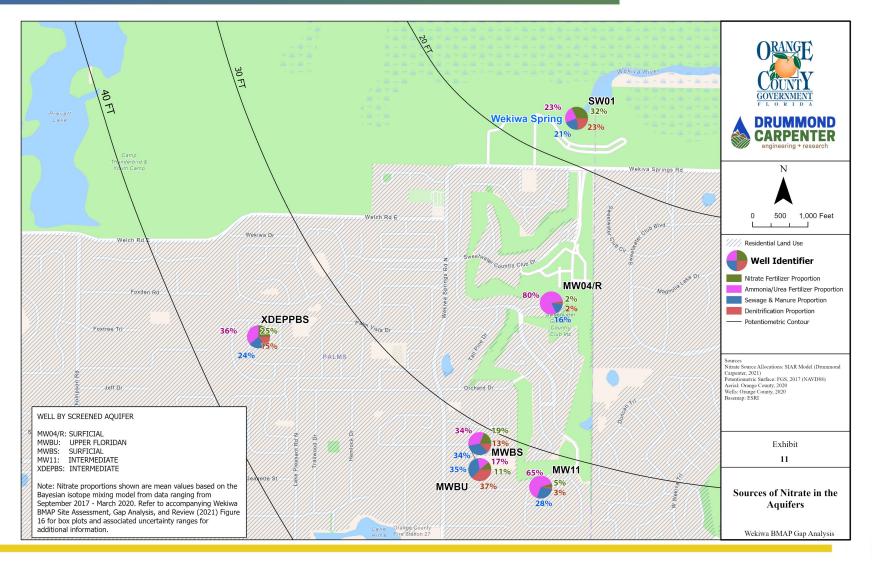
• Denitrification signal present.







Mixing model results can be viewed spatially.



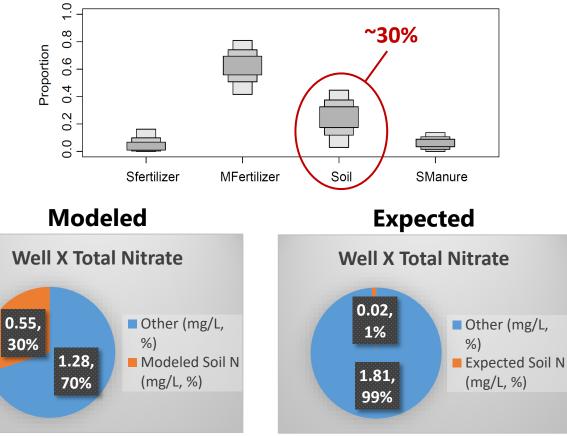


Springs Protection Assessment Using Nitrate Isotope Analysis CASE STUDY: WEKIWA SPRING

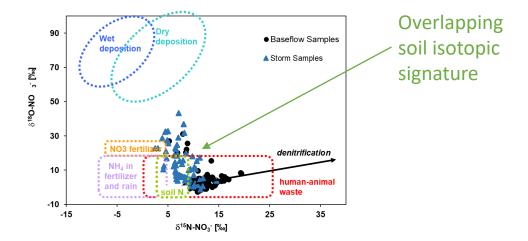


Real world 'check': do modeled proportions fall within expected or reasonable ranges based on knowledge of

land use, fertilizer practices, septic systems present, atmospheric N, background soil N concentrations?



Well X (Total Nitrate = 1.83 mg/L)



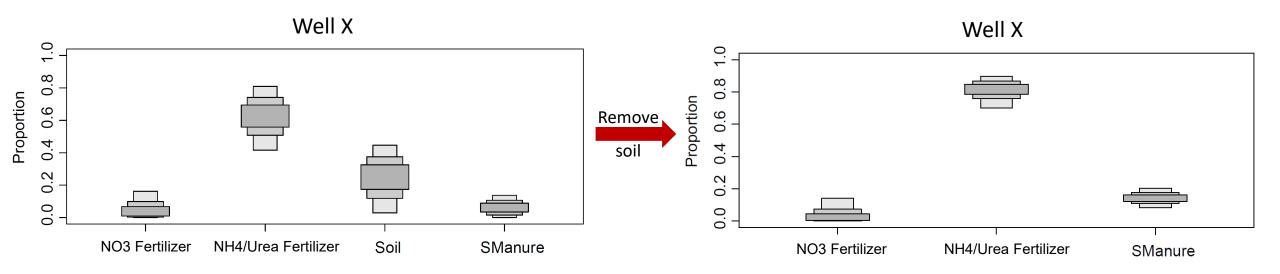
Overlapping isotopic signatures create uncertainty in model and overestimation of soil contribution





Real world 'check': do modeled proportions fall within expected or reasonable ranges based on knowledge of land use, fertilizer practices, septic systems present, atmospheric N, background soil N concentrations?

• Assuming soil-sourced nitrate is negligible reduces uncertainty in mixing model.





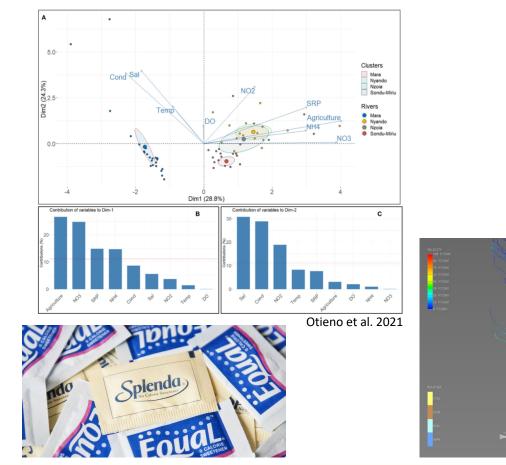


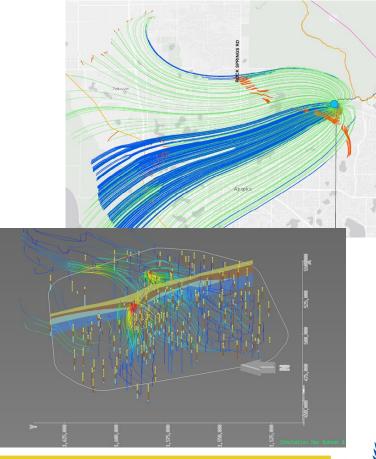
Multiple lines of evidence are helpful to compliment isotopic data.

- Previous studies
- Historical values
- Additional tracers
- Water quality data
- Transport models



Images courtesy of Google Images





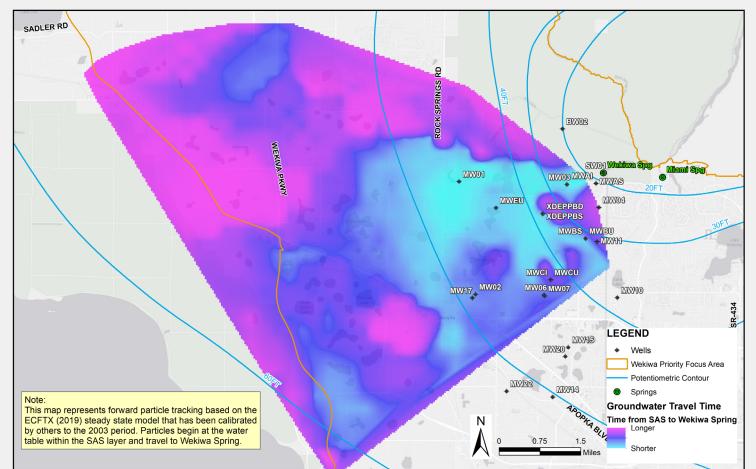


Springs Protection Assessment Using Nitrate Isotope Analysis OUTCOMES



Multiple lines of evidence can be used to build a better conceptual model.

- Identification of hotspots and vulnerable areas:
 - High recharge areas
 - Fast groundwater travel times
 - Elevated nitrate concentrations
 - Land use
- Management practices can be designed to target identified areas.





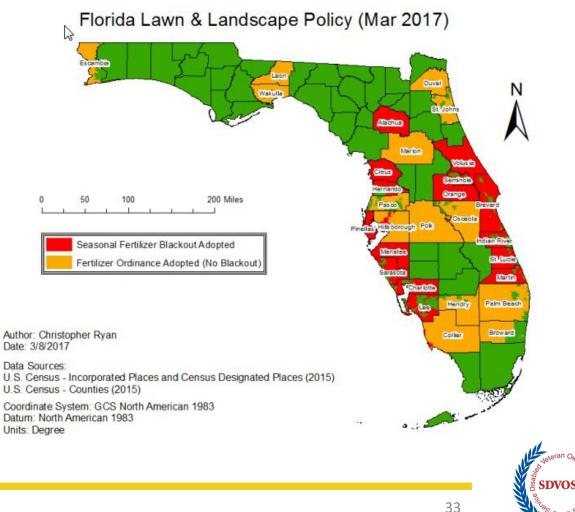
Springs Protection Assessment Using Nitrate Isotope Analysis OUTCOMES



Implement management practices to reduce nitrogen loading in vulnerable areas.

Fertilizer ordinances:

- Restrictions on fertilizer application rates
- Restrictions on fertilizer timing
- Mandates for fertilizers with slow-release Nitrogen
- Bans on fertilizer during wet season



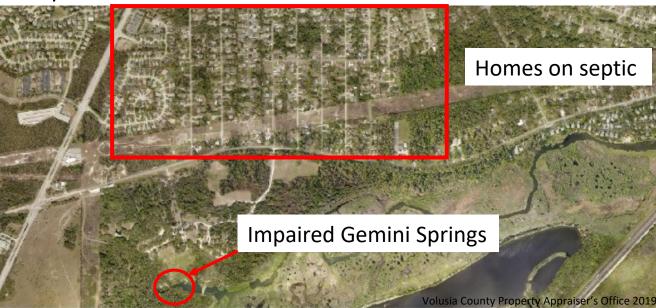
Springs Protection Assessment Using Nitrate Isotope Analysis OUTCOMES

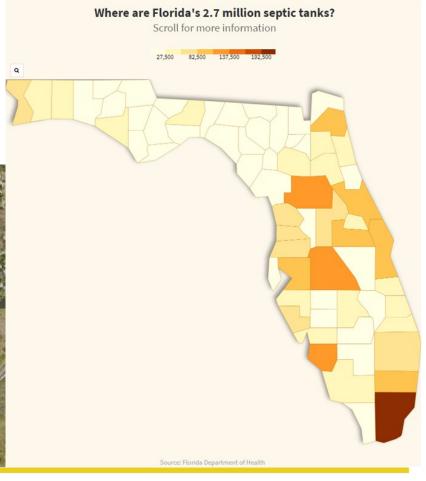


Implement management practices to reduce nitrogen loading in vulnerable areas.

Septic systems:

- Limit or prohibit permits for new septic
- Advanced treatment septic systems
- Septic-to-sewer retrofits









- Nitrate from multiple human-influenced sources contributes to eutrophication in Florida Springs.
- Dual Nitrate Isotope Analysis can help untangle the mix of sources observed in water samples.
- Mixing models further help to define source proportions to each sampling site.
- Mixing models also provide important estimates about denitrification.
- Understanding the sources and proportion from each help target solutions.

Rules to Follow

Restricted Season: June 1 - September 30

- No fertilizer containing nitrogen or phosphorus may be applied in Orange County from June 1 through September 30, unless you comply with one of the following:
 - Residential applicators must have annual proof of training via completion of the online Orange County Fertilizer Application Education Course for Citizens.
 - Commercial applicators must have a valid FDACS Limited Fertilizer Applicator's License or proof of completion of Green Industries Best Management Practices (GI-BMP) within the previous 3 years.

Press release from Orange County, September 27, 2021

Orange County Awarded Millions for Septic Conversion and Upgrades

Approximately \$41 million will go towards sewer conversion projects in Wekiwa Springs and Pine Hills



https://www.orangecountyfl.net/Environment/FertilizeResponsibly.aspx#.YYRCivnMKzU



THANK YOU!

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Special thanks to:

Orange County Environmental Protection Division





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